



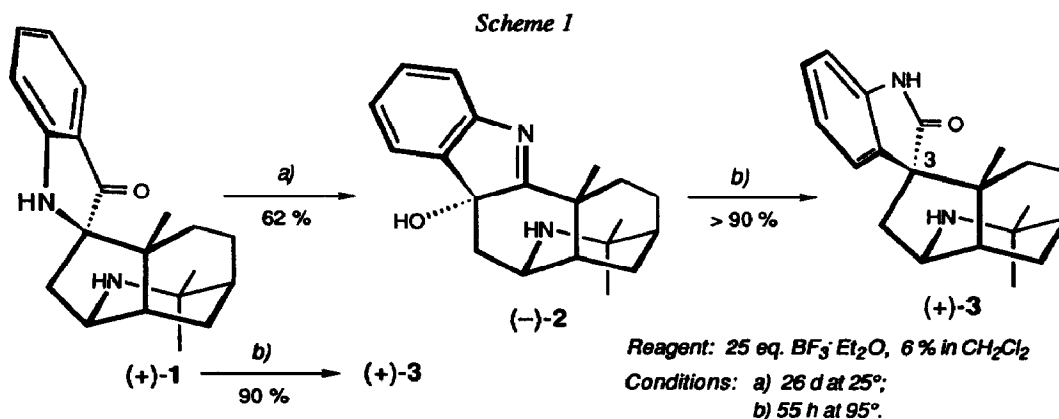
A Stereoselective Transformation of Pseudoindoxyls into Oxindoles in a Single Operation

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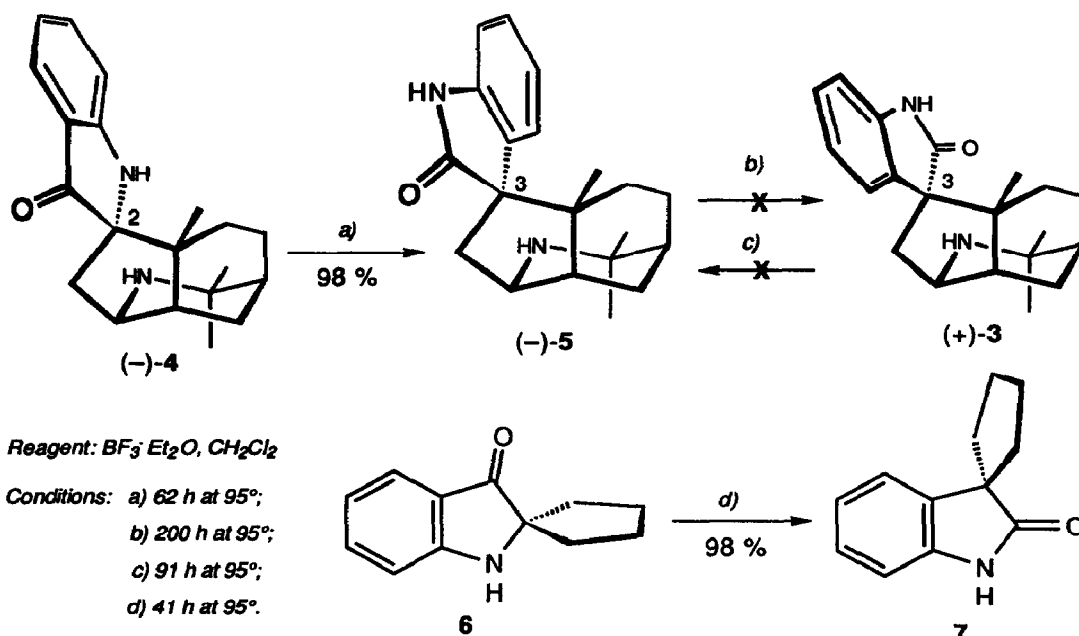
Abstract: The *Aristotelia* alkaloid (+)-aristolone (1), a spiro-pseudoindoxyl derivative, is transformed in over 90 % yield into the oxindole (+)-3-epitasmnine (3) upon treatment with hot $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in CH_2Cl_2 . This intriguing transformation possibly proceeds through the intermediate 3-hydroxyindolenine derivative (-)-serratoline (2) which could be isolated when the reaction was run under milder conditions. This rearrangement, for which there is little precedent, is highly stereoselective in that the lactam carbonyl group ends up on the same face of the molecule as the C=O-unit of the starting pseudoindoxyl. That this outcome is due to a kinetic control was demonstrated by showing that the epimeric starting material (-)-4 furnished exclusively the naturally occurring alkaloid (-)-tasmanine (5) under the same reaction conditions.

In connection with work concerned with the total synthesis and with oxidative transformations of certain *Aristotelia* alkaloids¹ we discovered that synthetic (+)-aristolone (1)² is slowly transformed into the isomeric compound (-)-serratoline (2) when treated with $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in CH_2Cl_2 at 25° (Scheme 1). In the crude reaction mixture 2 % of the constitutional isomer (+)-3-epitasmnine (3)^{2,3} could also be detected by means of ¹H-nmr spectroscopy (characteristic high-field singlet at 0.57 ppm due to the angular methyl group). Additional experiments under more stringent conditions (55 h at 95° in a sealed glass tube) showed that (-)-2 is transformed into (+)-3 in virtually quantitative yield, demonstrating that (-)-2 can serve as a precursor for the latter. Under the same conditions, pseudoindoxyl (+)-1 furnished the oxindole (+)-3 in a single operation.⁴



Additional experiments showed that under the above conditions an equilibrium between 1 and 2 is set up, which could be approached starting from both sides. Furthermore, it was shown that the slow formation of the oxindole 3 from either of these precursors is an irreversible process (3 was recovered absolutely unchanged after exposure to $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in CH_2Cl_2 at 95° for 8 days). A control experiment demonstrated that the spiro epimer (-)-tasmanine (5)³(Scheme 2) was also stable under these conditions, thus showing that the exclusive formation of 3 from 1 or 2 is not the result of a subsequent equilibration of the oxindoles 3 and 5. We corroborated this contention by verifying that the epimeric starting material (-)-2-epiaristotelone (4)² rearranges stereoselectively (> 95%) into (-)-tasmanine (5).

Scheme 2



To the best of our knowledge, the only precedent for the observed double 1,2-migration from carbon to carbon, accompanied by a concomitant 1,2-shift of a carbonyl group in the opposite direction, is represented by an analogous transformation of 2,2-diphenyl-1,2-dihydro-3*H*-indol-3-one into 3,3-diphenyl-1,3-dihydro-2*H*-indol-2-one, which was reported by *Witkop* and *Ek* in 1951.^{6,7} Nonetheless, this reaction seemingly is not restricted to that substrate and the *Aristolelia* alkaloids, since under the same conditions, tetrahydrocarbazole pseudoindoxyl (6)⁸ also rearranged to the corresponding oxindole 7⁹ in virtually quantitative yield.¹⁰ In this experiment, as well as during the transformation (-)-4 \rightarrow (-)-5, no intermediate 3-hydroxyindolenine derivative could be detected in the reaction mixture, even when the reaction was run under milder conditions (25°C , shorter reaction time). This negative evidence notwithstanding, the first step in the observed rearrangement sequence is most likely represented by a 1,2-shift of a C,C-bond involved in the set-up of the spiro center. In an asymmetric environment, such as in the case of (+)-1 and (-)-4, either of the bonds adjacent to the carbonyl group could migrate to furnish a hydroxyindolenine intermediate belonging to the regular series ((-)-2) or to the invert allo-series (8) (Scheme 3).^{2,13-15} At present, we are unable to decide whether pathway A or B is followed; A is electronically favored, but would lead to an intermediate (8), which is known to be much less stable than the alternative (-)-2.¹⁶ The next step might well involve either of the epoxides 9 or 11, as originally proposed by *Witkop*.^{6,17} These unstable intermediates can undergo alternative oxirane ring-opening reactions to furnish either their respective precursors 2 and 8, or the iminium salts I and II, respectively, both of which are expected to rearrange to the observed final product (+)-3. As an alternative, hydrated hydroxyindolenines, such as *cis*-diol 10, can be considered as possible reactive intermediates.^{18,19}

References and Notes

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- Although the presence of water under the given reaction conditions seems unlikely, this variant can not be ruled out since only catalytic amounts of H₂O would be required for the sequence **2** → **10** → **3**.

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